

Bis[2'-(methoxycarbonylethyl)-5'-tetrahydrofuryl]methane was obtained by the hydrogenation of bis[2'-(β -methoxycarbonylethyl)-5'-furyl]methane over Raney nickel at 100–110° C and an initial pressure of hydrogen of 200 atm with a yield of 70%. Bp 164–165° C (3 mm); n_D^{20} 1.4737; d_4^{20} 1.1119. Found, %: C 62.06, 62.30; H 8.48, 8.23; MR_D 82.95. Calculated for C₁₇H₂₈O₆, %: C 62.17; H 8.59%; MR_D 82.90.

REFERENCES

1. Dinelli and G. B. Marini, *Gazz. chim. ital.*, **67**, 312, 1937.
2. R. C. Ackman, W. H. Brown, and G. F. Wright, *J. Org. Chem.*, **20**, 1147, 1955.
3. L. L. J. Fauque, U. S. Patent 2681917, 1954; *C. A.*, **49**, 7600, 1955.

4. W. H. Brown and H. Sawatzky, *Can. J. Chem.*, **34**, 1147, 1956.

5. O. Moldenhauer, W. Irion, and O. H. Marwitz, German Federal Republic Patent 941062, 1956; RZhKh, 2157P, 1957.

6. L. Decsei, B. Zsador, and G. Reichmann, *Magyar. Kem. folyozat*, **69**, 328, 1964.

7. Y. Yahama and I. Hayshi, Japanese Patent 3696, 1956; *C. A.* **51**, 13930, 1957.

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NITROGEN-CONTAINING ORGANOSILICON COMPOUNDS

XIV. N-Morpholyl-, N-Thiamorpholyl-, N-Methylpiperazinyl-, and N-Perhydroazepinylalkylsilanes*

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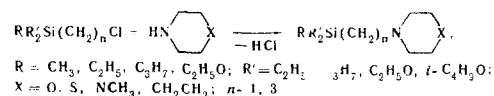
By the reaction of trialkylsilanes and dialkoxy- or di(chloroalkyl)alkylsilanes with morpholine, thiamorpholine, N-methylpiperazine, and hexamethyleneimine, 14 new organosilicon derivatives of these heterocycles have been synthesized. The addition of perhydroazepine to diethylmethylvinylsilane has been performed.

The usual method for obtaining morpholylsilanes [2] and piperazinylsilanes [3–7] containing the Si–N bond is the reaction of organylchlorosilanes with morpholine and piperazine, respectively. Their triphenylsilyl derivatives have been obtained by the reaction of the heterocycles mentioned with triphenylsilyl-lithium [8].

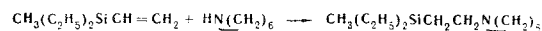
N-Piperazinylsilanes are also formed by the transamination of piperazine with alkylaminosilanes [9], dehydrocondensation with triorganylsilanes in the presence of potassium [10], and the thermal rearrangement of N-aziridinylsilanes [11]. N-Morpholinylmethylsilanes have been synthesized by the reaction of triorganylchloromethylsilanes with morpholine [12]. N-Piperazinylalkylsilanes containing the Si–C_n–N grouping have been obtained by the addition of piperazine to trialkylvinylsilanes in the presence of alkali metals [13] and also by the thermal rearrangement of aziridinylalkylsilanes [13, 14]. Triphenoxy-N-perhydroazepinyl-

silane is formed by the reaction of chlorotriphenoxysilane with hexamethylenediamine. Di-n-butoxy(methyl)-(N-perhydroazepinylmethyl)silane has been obtained by the reaction of hexamethyleneimine with di-n-butoxy-(chloromethyl)methylsilane [16]. Organosilicon derivatives of thiamorpholine were not previously known.

By heating trialkylsilanes, dialkoxyalkyl- and di(chloromethyl)alkylsilanes, and (3-chloropropyl)silanes with morpholine, thiamorpholine, N-methylpiperazine, and hexamethyleneimine (perhydroazepine) in toluene in the presence of triethylamine for 15–20 hr, we have obtained the corresponding alkyl-N-heterylsilanes with yields of 40–60%:



By the addition of hexamethyleneimine to diethylmethylvinylsilane in the presence of lithium, we have obtained diethylmethyl(2-N-perhydroazepinylethyl)silane:



The method of performing the syntheses was similar to that described previously [17]. The physicochemical constants, the analytical data, and the yields of the compounds synthesized are given in the table. A

*For part XIII, see [1].

N-Morpholyl-, N-Thiamorpholyl-, N-Methylpiperazinyl-, and N-Perhydroazepinylalkylsilanes ($\text{NC}_4\text{H}_8\text{O} =$ N-morpholyl, $\text{NC}_4\text{H}_8\text{S} =$ N-thiamorpholyl, $\text{NC}_4\text{H}_8\text{NCH}_3 =$ N-methylpiperazinyl, $\text{NC}_6\text{H}_{12} =$ N-perhydroazepinyl).

Compound	Bp, °C (pressure, mm)	n_D^{20}	d_4^{20}	$M\bar{R}_D$		Empirical formula	Found, %			Calculated, %			Yield %
				found	calculated		C	H	Si	C	H	Si	
$\text{CH}_3(\text{C}_2\text{H}_5)_2\text{SiCH}_2\text{NC}_4\text{H}_8\text{O}$	79—81.5 (3)	1.4623	0.9067	61.09	61.27	$\text{C}_{10}\text{H}_{23}\text{NOSi}$	59.64	11.87	13.81	59.64	11.51	13.95	65
$\text{CH}_3(i\text{-C}_4\text{H}_9\text{O})_2\text{SiCH}_2\text{NC}_4\text{H}_8\text{O}$	104—108 (2)	1.4402	0.9421	81.10	81.01	$\text{C}_{11}\text{H}_{31}\text{NO}_2\text{Si}$	57.37	10.39	9.71	58.08	10.79	9.70	41.9
$(\text{C}_2\text{H}_5)_3\text{SiCH}_2\text{NC}_4\text{H}_8\text{O}$	96.5—98 (4)	1.4680	0.9169	65.31	65.78	$\text{C}_{11}\text{H}_{25}\text{NOSi}$	61.10	12.01	12.20	61.33	11.70	13.04	62
$(\text{C}_3\text{H}_7)_3\text{SiCH}_2\text{NC}_4\text{H}_8\text{O}$	123 (4)	1.4621	0.8817	80.30	80.03	$\text{C}_{14}\text{H}_{31}\text{NOSi}$	65.22	12.28	—	65.30	12.13	10.91	40
$(\text{C}_2\text{H}_5\text{O})_3\text{SiCH}_2\text{NC}_4\text{H}_8\text{O}$	127.5—129 (15)	1.4339	1.0024	68.42	68.42	$\text{C}_{11}\text{H}_{25}\text{NO}_4\text{Si}$	50.54	9.54	10.84	50.16	9.57	10.66	69.2
$\text{CH}_3(\text{C}_2\text{H}_5)_2\text{Si}(\text{CH}_2)_3\text{NC}_4\text{H}_8\text{O}$	114—117 (6)	1.4640	0.8999	70.35	70.53	$\text{C}_{12}\text{H}_{27}\text{NOSi}$	62.99	12.07	11.95	62.82	11.86	12.24	55.4
$\text{CH}_3(\text{C}_2\text{H}_5\text{O})_2\text{Si}(\text{CH}_2)_3\text{NC}_4\text{H}_8\text{O}$	96 (0.5)	1.4460	0.9662	72.16	72.29	$\text{C}_{12}\text{H}_{27}\text{NO}_3\text{Si}$	54.43	10.34	11.06	55.13	10.41	10.74	41.5
$\text{CH}_3(\text{C}_2\text{H}_5\text{O})_2\text{Si}(\text{CH}_2)_3\text{NC}_4\text{H}_8\text{S}$	125—128 (1.5)	1.4741	0.9970	78.24	78.37	$\text{C}_{12}\text{H}_{27}\text{NO}_2\text{SSi}$	*	9.84	10.43	51.94	9.81	10.12	40
$\text{CH}_3(\text{C}_2\text{H}_5)_2\text{SiCH}_2\text{NC}_4\text{H}_8\text{NCH}_3$	110—114 (15)	1.4660	0.8722	68.05	67.94	$\text{C}_{11}\text{H}_{25}\text{N}_2\text{Si}$	61.35	12.20	12.65	61.61	12.22	13.10	37
$\text{CH}_3(\text{C}_2\text{H}_5)_2\text{Si}(\text{CH}_2)_3\text{NC}_4\text{H}_8\text{NCH}_3$	115.5—118 (3)	1.4663	0.8712	77.13	77.20	$\text{C}_{13}\text{H}_{30}\text{N}_2\text{Si}$	64.54	12.37	11.40	64.39	12.47	11.58	38
$\text{CH}_3(\text{C}_2\text{H}_5\text{O})_2\text{Si}(\text{CH}_2)_3\text{NC}_4\text{H}_8\text{NCH}_3$	124—126 (3)	1.4513	0.9327	78.96	79.28	$\text{C}_{13}\text{H}_{30}\text{N}_2\text{O}_2\text{Si}$	56.98	10.96	10.09	56.89	11.02	10.23	32
$\text{CH}_3(\text{C}_2\text{H}_5)_2\text{SiCH}_2\text{NC}_6\text{H}_{12}$	93—94 (4)	1.4675	0.8662	68.43	68.76	$\text{C}_{12}\text{H}_{27}\text{NSi}$	68.11	12.93	12.52	67.53	12.75	13.16	42
$\text{CH}_3(\text{C}_2\text{H}_5)_2\text{Si}(\text{CH}_2)_2\text{NC}_6\text{H}_{12}$	113 (4)	1.4713	0.8680	73.29	73.39	$\text{C}_{13}\text{H}_{29}\text{NSi}$	69.00	13.13	12.47	68.64	12.85	12.35	30
$\text{CH}_3(\text{C}_2\text{H}_5)_2\text{Si}(\text{CH}_2)_3\text{NC}_6\text{H}_{12}$	113—117 (3)	1.4698	0.8663	77.77	78.02	$\text{C}_{14}\text{H}_{31}\text{NSi}$	69.79	12.93	11.17	69.63	12.94	11.63	33

* Found, %: S 11.27. Calculated, %: S 11.5.

separate communication is devoted to the pharmacological characteristics of the diethoxymethyl derivatives [18].

REFERENCES

1. E. Lukevits and M. G. Voronkov, *ZhOKh*, **38**, 2325, 1968.
2. R. A. Pike and R. L. Schank, *J. Org. Chem.*, **27**, 2190, 1962.
3. L. W. Breed, W. J. Haggerty, and J. Harvey, *J. Org. Chem.*, **25**, 1804, 1960.
4. J. F. Klebe, *J. Polym. Sc.*, **A2**, 2673, 1964.
5. J. F. Klebe, US Patent 3207707, 1965; *C. A.*, **63**, 15004, 1965.
6. J. M. Maselli, US Dept. Com., Off. Tech. Serv., AD 274520, 1962; *C. A.*, **58**, 10311, 1963.
7. H. Pflieger, US Dept. Com., Off. Tech. Serv. AD 263736, 1961; *C. A.*, **58**, 2466, 1963.
8. H. Gilman and G. D. Lichtenwalter, *J. Am. Chem. Soc.*, **82**, 3319, 1960.
9. K. Lienhard and E. G. Rochow, *Angew. Chem.*, **75**, 980, 1963.
10. N. S. Nametkin, E. S. Vasil'eva, and V. N. Perchenko, *KhGS [Chemistry of Heterocyclic Compounds]*, **3**, 138, 1967.
11. N. S. Nametkin, V. N. Perchenko, and L. G. Batalova, *DAN*, **160**, 1087, 1965.
12. K. A. Andrianov and L. M. Volkova, *Izv. AN SSSR, OKhN*, **577**, 1957.
13. I. A. Grushevenko, Author's Abstract of Candidate's Thesis [in Russian], Moscow, 1967.
14. N. S. Nametkin, I. A. Grushevenko, and V. N. Perchenko, *DAN*, **162**, 347, 1965.
15. R. Schwarz and F. Weigel, *Z. anorg. allg. Chem.*, **268**, 291, 1952.
16. Z. I. Shabarova and M. G. Voronkov, *Izv. AN Latv. SSR, Ser. khim.*, **321**, 1967.
17. E. Lukevits, A. E. Pestunovich, and M. G. Voronkov, *KhGS [Chemistry of Heterocyclic Compounds]*, **3**, 559, 1967.
18. E. Lukevits, M. G. Voronkov, A. E. Pestunovich, A. A. Kimenis, S. Z. Gutberga, and Z. A. Atare, *Izv. AN Latv. SSR*, **93**, 1968.

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